

Tetrahedron 56 (2000) 557-561

Synthesis of the γ-Amino-β-hydroxy Acid of Hapalosin via an Asymmetric Dihydroxylation Route

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Received 29 October 1999; accepted 3 December 1999

Abstract—Starting from the allylic alcohol **3**, the epoxide **7** was prepared by asymmetric dihydroxylation of the allylic chloride followed by subsequent protection of the secondary hydroxy group. Opening of the oxirane with phenyl cuprate gave the triol **8** with a free hydroxy group. Mitsunobu reaction of **8** with diphenylphosphoryl azide led to the azide **9**. Simple functional group manipulations delivered the acid **2** in further five steps. © 2000 Elsevier Science Ltd. All rights reserved.

Introduction

The cyclic depsipeptide hapalosin (1) was found to reverse multidrug resistance^{1,2} (MDR) in tumor cells.³ Common MDR is caused by overexpression of the *MDR1* gene that encodes for a 170 kDa P-glycoprotein. Structurally, hapalosin consists of three subunits, a β -hydroxy acid **A**, an α -hydroxy acid **B**, and a γ -amino- β -hydroxy acid **C**. With the modular structure, this molecule presents itself for analog synthesis in a parallel fashion.⁴ Due to the biological activity, syntheses of the natural product⁵⁻¹⁰ and of some analogs^{10,11} have been reported.



While the synthesis and availability of subunits of type **A** and **B** are no problem, the (3R,4S)-4-*N*-methylamino-3-hy-

droxy-5-phenylpentanoic acid requires a higher synthetic effort. Typically, amino acid **C** is prepared by elongation of *N*-protected phenylalanine in a sequence of about four to seven steps. Moreover, we recently disclosed the synthesis of acid **2** using the tactical combination of an aldol reaction and a Curtius rearrangement as key steps.¹² However, this and the other known syntheses do not allow for an easy modification or replacement of the aryl group. We therefore devised another route towards **2** that introduces the aryl group through the opening of an epoxide. The stereo-chemical information was established by a Sharpless asymmetric dihydroxylation reaction (ADH).¹³

Results

The synthesis started from the allylic alcohol **3**, which is available from benzyloxypropanol by an oxidation, Wittig–Horner and reduction sequence.^{14–16} Chlorination¹⁷ with *N*-chlorosuccinimide and dimethyl sulfide provided allyl chloride **4** in high yield (Scheme 1). The asymmetric dihydroxylation was performed under standard conditions¹⁸



Scheme 1.

Keywords: hapalosin; amino acid; epoxides; hydroxylation.

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Scheme 2.

using $(DHQD)_2PHAL$ as chiral ligand, which furnished the (2S,3R)-diol **5**. Treatment of the chlorodiol **5** with pulverized sodium hydroxide led to the epoxy alcohol **6**. Despite the presence of the epoxide, protection of the hydroxyl group as its MOM ether proceeded uneventually to give the key building block **7**.

The stage was now set for the introduction of the phenyl group. Thus, the higher order cyanocuprate,¹⁹ prepared from phenyllithium (2 equiv.) and copper cyanide added smoothly to the terminal carbon of the epoxide **7** affording the di-protected triol **8** in good yield (Scheme 2). The alcohol **8** was used to determine the enantioselectivity of the dihydroxylation reaction. It was converted to the corresponding Mosher esters using both Mosher acids.^{20,21} An equimolar mixture of the Mosher esters could be separated on a DAICEL ODH column [250×4.6 mm, 2% *i*PrOH in heptane, flow 0.9 ml min⁻¹, retention times 11.99 and 13.40 min, 254 nm]. Injecting the Mosher ester derived from the (*S*)-acid (**8MS**) gave basically only one peak (13.40 min). The ratio from integration indicated an ee-value of 96%.

In the next step the nitrogen function was introduced through a Mitsunobu reaction²² with diphenyl-phosphoryl azide¹⁰ in the presence of triphenylphosphine and diethyl azodicarboxylate to furnish the azide 9. This substitution reaction went in high yield but was accompanied by the elimination product 10, which could be separated by chromatography.

After reduction of the azide with $LiAlH_4$ the amine **11** was obtained (Scheme 3). The free amine could be converted to the corresponding *N*-Boc derivative **12** (Boc₂O, NEt₃, trace

DMAP, THF) and then by methylation (1.3 equiv. NaH, 2.0 equiv. MeI, DMF) to the protected amino alcohol **13**. Reducitve removal of the benzyl group gave the known primary alcohol **14**.¹² The measured optical rotation of **14** $[\alpha]_D^{25} = -16.0$ (c=0.9 in CHCl₃) {Ref. 12 $[\alpha]_D^{25} = -17.8$ (c=0.9 in CHCl₃)} was proof of the high enantioselectivity in the ADH reaction. To complete the synthesis, the primary alcohol **14** was oxidized to the acid **2** with sodium hypochlorite and TEMPO as described.¹²

In summary, we prepared the chiral oxirane 7 by an asymmetric dihydroxylation of the allylic chloride 4 and illustrated the use of this polyfunctional C-5 building block by its conversion to the 4-amino-3-hydroxy-5-phenyl-pentanoic acid 2.

Experimental

General

¹H and ¹³C NMR: Bruker AC 250; all spectra were recorded in CDCl₃. Optical rotations: JASCO P-1020 polarimeter. IR: JASCO FT/IR-430 spectrometer, EI-MS: AMD Intectra GmbH AMD 402. HPLC: Hewlett Packard HP 1100. Flash chromatography: J. T. Baker silica gel 30–60 μm. Thin-layer chromatography: Merck Si 60 F₂₅₄. Solvents were distilled prior to use; petroleum ether with a boiling range of 35–65°C was used.

Benzyl (3*E***)-5-chloro-3-pentenyl ether (4).** To a solution of NCS (15.4 g, 115 mmol) in dry CH_2Cl_2 (300 ml) was added DMS (8.5 ml, 115 mmol) at 0°C. After stirring for 1 h at 0°C the mixture was cooled to -20°C and a solution



of the allylic alcohol 3 (11.1 g, 57.7 mmol) in CH_2Cl_2 (100 ml) was added dropwise over a period of 30 min. After stirring for 30 min at 0°C the mixture was allowed to warm slowly to room temperature and stirred at this temperature for 1 h. Water (150 ml) was added, the two layers were separated, and the CH₂Cl₂ layer was washed with brine (2×150 ml), dried (Na₂SO₄), filtered and concentrated in vacuo. Purification of the residue by flash chromatography on silica gel (petroleum ether/ethyl acetate 96:4 and 90:10) gave 4 (11.5 g, 94%) as a colorless oil. TLC (petroleum ether/ethyl acetate, 95:5): $R_f=0.48$; IR (neat): 1495, 1453, 1363, 1252, 1204, 1102, 1028, 967, 738, 698 cm⁻¹; ¹H NMR (250 MHz, CDCl₃): δ =2.39 (dt, J=6.5, 6.5 Hz, 2H), 3.50 (t, J=6.6 Hz, 2H, CH₂OBn), 4.04 (d, J=6.7 Hz, 2H, CH₂Cl), 4.50 (s, 2H, CH₂Ph), 5.60–5.90 (m, 2H, olefinic), 7.26–7.35 (m, 5H); ¹³C NMR $(62.9 \text{ MHz}, \text{CDCl}_3): \delta = 32.6, 45.2, 69.3, 73.0, 127.7, 128.4,$ 132.3, 138.0; HRMS (EI) calcd for C₁₂H₁₅ClO 210.08113, found 210.07991.

5-O-Benzyl-1-chloro-1,4-dideoxy-D-threo-pentitol (5). A solution of (DHQD)₂PHAL (370 mg, 1 mol%), K₂OsO₂ (OH)₄ (87 mg, 0.5 mol%), K₃[Fe(CN)₆] (47.0 g, 142 mmol), K₂CO₃ (20.0 g, 142 mmol), NaHCO₃ (12 g, 142 mmol) and $CH_3SO_2NH_2$ (4.50 g, 47.5 mmol) in *t*-BuOH/H₂O (1:1, 500 ml) was cooled to 0°C and treated with the allylic halide 4 (10.0 g, 47.5 mmol). The mixture was stirred vigorously for 24 h at 0°C. Na₂S₂O₅ (45 g, 237 mmol) was added carefully, and stirring was continued for 1 h at room temperature. The layers were separated and the aqueous layer was extracted with ethyl acetate (150 ml). The combined organic layers were washed with a 1N KOH (150 ml), 5% aqueous HCl (150 ml) and brine (150 ml), dried (MgSO₄), filtered and concentrated under reduced pressure. The residue was purified by flash chromatography on silica gel (petroleum ether/ethyl acetate, 1:1 and 2:3) to give 5 (11.1 g, 95%) as a colorless solid. $[\alpha]_{D}^{29} = -13.0$ (c 2.0, CH₂Cl₂); TLC (petroleum ether/ethyl acetate, 6:4): $R_f=0.44$; IR (neat): 1453, 1366, 1252, 1094, 849, 742, 6999, 612 cm⁻¹; ¹H NMR $(250 \text{ MHz}, \text{ CDCl}_3)$: $\delta = 1.73 - 1.82$, 1.91 - 2.08 (2 m, 2H). 2.9 (br s, 1H, OH), 3.30 (br s, 1H, OH), 3.56-3.77 (m, 5H, CH₂Cl₂CH₂OBn, CH) 3.97-4.02 (m, 1H, CH), 4.53 (s, 2H, CH₂Ph), 7.26–7.40 (m, 5H); ¹³C NMR (62.9 MHz, CDCl₃): δ=33.2, 45.9, 68.5, 70.6, 73.5, 74.0, 127.8, 128.6, 137.6.

4,5-Anhydro-1-O-benzyl-2-deoxy-D-threo-pentitol (6). To a cooled (0°C) solution of chlorodiol 5 (5.0 g, 20.5 mmol) in THF (30 ml) was added pulverized NaOH (1.64 g, 41.0 mmol). After stirring for 2 h at 0°C the mixture was diluted with diethyl ether (100 ml). Water (40 ml) was added, the layers were separated, and the organic layer was washed with saturated aqueous NH₄Cl (50 ml), brine (50 ml), dried (MgSO₄), filtered and concentrated in vacuo to give **6** (3.8 g, 89%) as a colorless oil. $[\alpha]_D^{28} = -2.8$ (c 1.0, CH₂Cl₂); TLC (petroleum ether/ethyl acetate, 6:4): $R_{\rm f}$ =0.44; ¹H NMR (250 MHz, CDCl₃): δ =1.82-1.93 (m, 2H), 2.59–2.61 (m, 1H, OH), 2.70–2.73, 2.75–2.79 (2m, 2H, CH₂O), 2.98–3.02 (m, 1H, CH), 3.63–3.79 (m, 3H, CH, CH₂OBn), 4.51 (d, J=1.5 Hz, 2H, CH₂Ph), 7.26–7.37 (m, 5H); 13 C NMR (62.9 MHz, CDCl₃): δ =33.9, 44.6, 55.1, 67.6, 69.9, 73.3, 127.7, 128.5, 138.0; HRMS (EI) calcd for C₁₂H₁₆O₃ 208.10993, found 208. 10051.

4,5-Anhydro-1-O-benzyl-2-deoxy-3-O-(methoxymethyl)-**D-threo-pentitol** (7). To a cooled $(0^{\circ}C)$ solution of epoxy alcohol 6 (4.60 g, 22.3 mmol) and diisopropylethylamine (11.4 ml, 66.8 mmol) in dry CH₂Cl₂ (6 ml) was added dropwise chloromethyl methylether (3.9 ml, 44.6 mmol). The reaction mixture was stirred at 0°C for 1 h and then at room temperature for 2 h. Water (60 ml) and CH₂Cl₂ (100 ml) were added, the two layers separated, and the aqueous layer was extracted with CH_2Cl_2 (2×60 ml). The combined CH₂Cl₂ layers were washed with 1 N HCl (40 ml), water (80 ml), dried (Na₂SO₄), filtered and concentrated in vacuo. Purification of the residue by flash chromatography on silica gel (petroleum ether/ethyl acetate, 80:20 and 75:25) gave 7 (5.0 g, 89%) as a colorless oil. $[\alpha]_{D}^{28} = +40.9$ (c 1.0, CH₂Cl₂); TLC (petroleum ether/ethyl acetate, 8:2): R_f=0.38; IR (neat); 1496, 1454, 1404, 1364, 1242, 1208, 1155, 1102, 1032, 919, 851, 813, 740, 699, 610 cm^{-1} ; ¹H NMR (250 MHz, CDCl₃): δ =1.86–1.94 (m, 2H), 2.51–2.55, 2.75–2.77 (2m, 2H, CH₂O), 2.99–3.04 (m, 1H, CH), 3.37 (s, 3H, CH₃), 3.44–3.52 (m, 1H, CH) 3.60 (t, J=6.2 Hz, 2H, CH₂OBn), 4.49 (s, 2H, CH₂Ph), 4.66, 4.84 (2 d, J=6.25 Hz, 1H each, CH₂OMe), 7.27–7.37 (m, 5H); ¹³C NMR (62.9 MHz, CDCl₃): δ =32.6, 43.9, 54.6, 55.7, 66.2, 73.1, 75.5, 95.8, 127.7, 128.4, 138.3; HRMS (EI) calcd for C₁₂H₁₅O₃ (M-CH₂OCH₃) 207.10211, found 207.10007.

5-O-Benzyl-1,4-dideoxy-3-O-(methoxymethyl)-1-phenyl-**D-threo-pentitol** (8). To a cooled $(-78^{\circ}C)$ suspension of copper cyanide (2.44 g, 27.2 mmol) in dry THF (60 ml) was added dropwise phenyllithium [30.2 ml, 1.8 M solution in diethylether/(30%) cyclohexane, 54.4 mmol]. The mixture was allowed to warm to -20° C for complete dissolution and was then recooled to -78° C. Now the epoxide 7 (4.90 g, 19.4 mmol) was added neat slowly via syringe. After warming to -20° C stirring was continued at this temperature for 8 h. The reaction was guenched with a mixture of 10% concentrated aqueous NH₃ in saturated aqueous NH₄Cl (60 ml) and allowed to warm to room temperature. Ether (60 ml) was added, the layers were separated followed by extracting the aqueous layer with ether (2×100 ml). The combined organic layers were washed with brine, dried (Na₂SO₄) and filtered. The solvent was removed in vacuo and the residue purified by flash chromatography (petroleum ether/ethyl acetate, gradient: 8:2, 7:3 and 6:4) to give 8 (6.04 g, 94%) as a light yellow oil. $[\alpha]_{D}^{25} = +15.3$ (c 0.51, CH₂Cl₂); TLC (petroleum ether/ethyl acetate, 8:2): $R_{\rm f}$ =0.17; IR (neat): 1603, 1495, 1454, 1364, 1208, 1100, 1030, 918, 743, 699, 606 cm^{-1} ; ¹H NMR (250 MHz, CDCl₃): δ =1.82–1.92, 1.97–2.04 (2m, 2H), 2.70–2.91 (m, 3H, CH₂Ph, OH), 2.85–2.92 (m, 2H, CH₂Ph), 3.40 (s, 3H, CH₃), 3.58–3.70 (m, 3H, CH₂OBn, CH), 3.76–3.83 (m, 1H, CH), 4.48 (d, J=1.9 Hz, 2H, CH₂Ph), 4.68–4.69 (m, 2H, CH₂OMe), 7.18–7.27, 7.28–7.38 (2m, 10H); 13 C NMR $(62.9 \text{ MHz}, \text{CDCl}_3); \delta = 31.6 39.6, 56.0, 56.0, 66.4, 73.1,$ 74.1, 76.6, 97.3, 126.3, 127.7, 128.5, 138.2, 139.0; HRMS (EI) calcd for C₁₉H₂₃O₃ (M–OCH₃) 299.16471, found 299. 15907.

Preparation of the Mosher esters from the alcohol 8. To a solution of (*S*)-MTPA (24 mg, 0.10 mmol) and DMF (7 mg, 0.1 mmol) in hexane (4 ml) was added dropwise $(COCl)_2$ (54.6 mg, 0.43 mmol) at room temperature. A white

precipitate formed immediately. After stirring for 1 h the precipitate had disappeared and some oily drops resulted. The hexane solution was separated and concentrated under reduced pressure. A solution of the di-protected triol **8** (25 mg, 0.076 mmol), DMAP (5 mg, 0.04 mmol) and NEt₃ (23 mg, 0.23 mmol) in CHCl₃ (1 ml) was added to the residue at room temperature. After 2 h CH₂Cl₂ (5 ml) was added and the solution washed with 1 M KHSO₄ (5 ml) and brine (5 ml), dried (MgSO₄), filtered and concentrated in vacuo. The residue was purified by flash chromatography on silica gel (petroleum ether/ethyl acetate, 8:2) to give the Mosher ester (33 mg, 80%) as a colorless oil. TLC (petroleum ether/ethyl acetate, 8:2); $R_{\rm f}$ =0.61.

Mosher ester derived from the (*S*)-acid: ¹H NMR (250 MHz, CDCl₃): δ =1.64–1.75 (m, 2H, CH₂), 2.75 (dd, *J*=14.4, 10.3 Hz, 1H, CH₂Ph), 3.11 (dd, *J*=14.4, 3.3 Hz, 1H, CH₂Ph), 3.19 (d, *J*=1.2 Hz, 3H, OCH₃), 3.34 (s, 3H, CH₃ (MOM)), 3.38–3.53 (m, 2H, CH₂OBn), 3.81–3.88 (m, 1H, CHOMOM), 4.39 (d, *J*=1.5 Hz, 2H, CH₂Ph), 4.57–4.63 (m, 2H, CH₂OMe), 5.49–5.54 (m, 1H, CH), 7.02–7.61 (m, 15H); ¹³C NMR (62.9 MHz. CDCl₃): δ =30.2, 34.6, 55.3, 56.1, 66.4, 73.0, 75.5, 78.1, 97.6, 125.6, 126.7, 127.7, 128.3, 128.6, 129.4, 132.3, 137.3, 138.4, 166.2.

Mosher ester derived from the (*R*)-acid: ¹H NMR (250 MHz, CDCl₃): δ =1.73–1.82, 1.83–1.91 (2m, 2H, CH₂), 2.73 (dd, *J*=14.4, 9.9 Hz, 1H, CH₂Ph), 3.04 (dd, *J*=14.4, 3.4 Hz, 1H, CH₂Ph), 3.23 (d, *J*=1.1 Hz, 3H, OCH₃), 3.34 (s, 3H, CH₃ (MOM)), 3.45–3.56 (m, 2H, CH₂OBn), 3.82–3.89 (m, 1H, CHOMOM), 4.41 (d, *J*=1.3 Hz, 2H, CH₂Ph), 4.60–4.64 (m, 2H, CH₂OMe), 5.46–5.53 (m, 1H, CH), 7.00–7.03, 7.07–7.12, 7.15–7.43 (3m, 15H); ¹³C NMR (62.9 MHz, CDCl₃): δ =30.8, 35.1 55.4, 56.0, 66.2, 73.1, 75.7, 77.8, 97.5, 125.6, 126.6. 127.5, 127.7, 128.3, 128.5, 129.3, 132.0, 137.1, 138.4, 166.1.

2-Azido-5-O-benzyl-1,2,4-trideoxy-3-O-(methoxymethyl)-1-phenyl-D-threo-pentitol (9). To a solution of the di-protected triol 7 (6.90 g, 20.9 mmol) and triphenylphosphine 6.0 g, 23 mmol) in dry THF (60 ml) at 0°C was added dropwise diethyl azodicarboxylate (DEAD, 4.0 g, 23 mmol) and subsequently diphenylphosphoryl azide (DPPA, 6.0 g, 21.8 mmol). After stirring for 20 min at 0°C the solution was allowed to stir at room temperature for 3 h. The reaction solution was partitioned between diethyl ether (100 ml) and water (80 ml), the two layers were separated, and the organic layer was washed with brine, dried (MgSO₄), filtered and concentrated in vacuo. The residue was purified by flash chromatography on silica gel (CH₂Cl₂ as eluent) to give 9 (6.54 g, 88%) as a colorless oil and the by-product **10** (0.55 g, 8%). $[\alpha]_D^{29} = +18.5$ (c 0.504, CH_2Cl_2 ; TLC (CH_2Cl_2): $R_f=0.53$; IR (neat): 2170, 2120, 1603, 1590, 1495, 1454, 1365, 1271, 1206, 1184, 1154, 1101, 1030, 965, 918, 739, 700, 618 cm⁻¹; ¹H NMR (250 MHz, CDCl₃):δ=1.89-1.96 (m, 2H), 2.64-2.85 (m, 2H, CH₂Ph), 3.39 (s, 3H CH₃), 3.59-3.66 (m, 2H, CH₂OBn), 3.81–3.91 (m, 2H, 2CH), 4.56–4.63 (m, 2H, CH₂Ph), 4.66 (q, 2H, CH₂OMe), 7.20–7.27, 7.28–7.39 (2m, 10H); ¹³C NMR (62.9 MHz, CDCl₃); δ =30.5, 36.8, 56.0, 66.4, 67.2, 73.1, 97.0, 126.8, 127.7, 128.5, 129.2, 130.1, 138.0, 138.3.

2-Amino-5-O-benzyl-1,2,4-trideoxy-3-O-(methoxymethyl)-**1-phenyl-D-threo-pentitol** (11). To a suspension of $LiAlH_4$ (1.1 g, 28.8 mmol) in dry diethyl ether (80 ml) was added dropwise a solution of the azide 9 (5.1 g, 14.4 mmol) in dry diethyl ether (10 ml) at room temperature. The reaction mixture was stirred for 3 h at room temperature. After cooling to 0°C wet diethyl ether (80 ml) was added followed by the dropwise addition of water (60 ml). The aqueous layer was removed and then extracted with ether (300 ml). The combined ether layers were washed with brine $(2 \times 100 \text{ ml})$, dried (MgSO₄), filtered and evaporated in vacuo. The residue was purified by flash chromatography on silica gel (50% petroleum ether in ethyl acetate and 20% methyl alcohol in ethyl acetate as eluents) to give 11 (4.1 g, 86%) as a colorless oil. $[\alpha]_{D}^{25} = +9.7$ (c 0.45, CH₂Cl₂); TLC (petroleum ether/ethyl acetate, 1:1): $R_f=0.1$; IR (neat): 3377 (w), 1603, 1495, 1454, 1364, 1207, 1151, 1098, 10.35, 917, 839, 740, 700, 607, 536 cm⁻¹; ¹H NMR (250 MHz, CDCl₃): $\delta = 1.43$ (br s, 2H, NH₂), 1.90–1.97 (m, 2H), 2.40-2.49, 2.83-2.89 (2m, 2H, CH₂Ph), 3.10-3.24 (m, 1H, CH), 3.37 (s, 3H, CH₃), 3.59–3.70 (m, 2H, CH₂OBn), 3.71-3.76 (m, 1H, CHOMOM), 4.51 (s, 2H, CH₂Ph), 4.63-4.70 (m, 2H, CH₂OMe), 7.17–7.26, 7.27–7.37 (2m, 10H); ¹³C NMR (62.9 MHz, CDCl₃):δ=30.2, 39.7 55.6, 55.8, 66.9, 73.1, 79.6, 96.8, 126.3, 127.6, 128.4, 129.2, 138.5, 139.6; HRMS (EI) calcd for C₂₀H₂₇NO₃ 329.19908, found 329.20495.

5-O-Benzyl-2-[(tert-butoxycarbonyl)amino]-1,2,4trideoxy-3-O-(methoxymethyl)-1-phenyl-D-threo-pentitol (12). To a cooled $(0^{\circ}C)$ solution of the amine 11 (4.2 g, 12.4 mmol) in dry tetrahydrofuran (40 ml) were added 16 mmol), triethylamine (Boc)₂O (3.5 g, (1.5 g, 14.8 mmol) and dimethylaminopyridine (DMAP, 0.3 g, 2.5 mmol). The solution was stirred for 2 h at room temperature, then diluted with ether (120 ml), washed successively with a 1 M aqueous KHSO₄ solution (40 ml), water (40 ml), a saturated aqueous NaHCO₃ solution (40 ml), water (40 ml) and brine. The organic layer was dried (Na_2SO_4), filtered and concentrated in vacuo. Purification of the residue by flash chromatography on silica gel (20% and 30% ethyl acetate in petroleum ether as eluents) gave 12 (4.9 g, 92%). $[\alpha]_{\rm p}^{26} = -18.0$ (c 0.5, CH₂Cl₂); TLC (petroleum ether/ethyl acetate, 8:2): $R_f=0.43$; IR (neat): 3364 (s), 1711 1604, 1498, 1454, 1365, 1245, 1171, 1100, 1034, 918, 851, 739, 699, 607, 463 cm^{-1} ; ¹H NMR $(250 \text{ MHz}, \text{ CDCl}_3): \delta = 1.31 \text{ (s, 9H)}, 1.83 - 1.91 \text{ (m, 2H)},$ 2.61-2.70 (m, 1H, CH₂Ph), 2.91 (dd, J=13.9, 4.8 Hz, 1H, CH₂Ph), 3.40 (s, 3H, CH₃), 3.51-3.65 (m, 2H, CH₂OBn), 3.76-3.82 (m, 1H, CH), 3.91-4.00 (m, 1H, CH), 4.50 (m, 2H, CH₂Ph), 4.53-4.61 (m, 2H, CH₂OMe), 5.08 (br d, J=8.5, 1H, NH), 7.14–7.26, 7.27–7.41 (2m, 10H); ¹³C NMR (62.9 MHz, CDCl₃): δ =28.3, 32.2, 36.2, 54.3, 55.9. 66.7, 73.1, 79.0, 97.3, 126.2, 127.8, 128.4, 129.3, 138.3, 138.6, 155.4.

5-O-Benzyl-2-[(*tert***-butoxycarbonyl)(methyl)amino]-1,2,4-trideoxy-3-O-(methoxymethyl)-1-phenyl-D-threopentitol (13).** To a cooled (0°C) solution of **12** (4.6 g, 10.8 mmol) and MeI (3.1 g, 21.5 mmol) in dry DMF (40 ml) was added NaH (0.56 g, 14 mmol, 60% in mineral oil). After complete addition, the suspension was allowed to warm to room temperature and stirred at this temperature for

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18 h. The reaction was quenched at 0°C with saturated aqueous NH₄Cl (25 ml). The mixture was partitioned between ethyl acetate (80 ml) and water (20 ml). The aqueous layer was extracted with ethyl acetate $(2 \times 80 \text{ ml})$. Then the combined organic layers were washed with water (3×40 ml), dried (Na₂SO₄), filtered and concentrated in vacuo. The residue was purified by flash chromatography on silica gel (30% ethyl acetate in petroleum ether) to give **13** (4.38 g, 91%) as a colorless oil. $[\alpha]_D^{25} = -26.2$ (c 0.578, CH₂Cl₂); TLC (petroleum ether/ethyl acetate, 8:2); $R_{\rm f}$ =0.43; IR (neat): 1738, 1693, 1604, 1585, 1496, 1479, 1454, 1391, 1365, 1321, 1244, 1169, 1151, 1034, 959, 919, 874, 824, 772, 738, 699, 607, 544, 520, 464 cm⁻¹; ¹H NMR (250 MHz, CDCl₃): δ (two rotamers)=1.19 (br, s, 9H, minor), 1.29 (s, 9H, major), 1.69-1.82, 1.93-2.05 (2m, 2H), 2.53 (s, 3H, minor, NCH₃), 2.68 (br s, 3H, major, NCH₃), 2.70–2.85 (m, 1H, CH₂Ph), 3.11–3.21 (m, 1H, CH), 4.18–4.22 (m, 1H, CH), 4.49 (s, 2H, CH₂Ph), 4.65– 4.78 (m, 2H, CH₂OMe), 7.12–7.26, 7.27–7.33 (2m, 10H); ¹³C NMR (62.9 MHz, CDCl₃): δ =28.3, 32.2, 32.4, 34.5, 56.4, 66.4, 66.8, 73.2, 79.3, 97.2, 126.1, 126.3, 127.7, 128.3, 129.2, 138.5, 139.2, 155.7; HRMS (EI) calcd for C₂₆H₃₇NO₅ 443.26715, found 443.27049.

2-[(tert-Butoxycarbonyl)(methyl)amino]-1,2,4-trideoxy-3-O-(methoxymethyl)-1-phenyl-D-threo-pentitol (14). A solution of the protected amine 13 (4.3 g, 9.7 mmol) in ethyl acetate (100 ml) was hydrogenated over 10% Pd/C (1.4 g) at room temperature and ordinary pressure for 5 h. After removal of the catalyst, the solution was concentrated in vacuo to leave an oil, which was purified by flash chromatography on silica gel (30% ethyl acetate in petroleum ether) to give 14 (3.3 g, 96%). $[\alpha]_{D}^{25} = -16.0$ (c 0.9, CHCl₃); TLC (petroleum ether/ethyl acetate, 3:7): $R_{\rm f}$ =0.42; IR (neat): 1738, 1686, 1604, 1496, 1480, 1454, 1392, 1366, 1324, 1249, 1217, 1153, 1034, 958, 920, 871, 824, 772, 751, 701, 545, 521, 464 cm⁻¹; ¹H NMR (250 MHz, CDCl₃): δ (two rotamers)=1.16, 1.24 (2 s, 3H, CH₃), 1.52–1.66, 1.82–1.98 (2m, 2H), 2.25 (br s, 1H, OH), 2.49 (s, 3H, minor NCH₃), 2.62 (s, 3H, major, NCH₃), 2.70-2.90 (complex, 1H, CH₂Ph), 3.01-3.14 (m, 1H, CH₂Ph), 3.38 (s, 3H, minor, CH₃), 3.41 (s, 3H, major, CH₃), 3.62-3.84 (m, 3H, CH₂(OH), CH), 4.20 (br s, 1H, CH), 4.64–4.76 (m, 2H, CH₂OMe), 7.07-7.23 (m, 5H); ¹³C NMR (62.9 MHz, CDCl₃): δ (two rotamers)=28.1 28.3, 34.1, 34.3, 34.6, 56.2, 56.4, 58.8, 78.3, 79.6, 79.8, 97.4, 97.7, 126.1, 126.3, 128.9, 129.0, 138.8, 155.5, 156.0; HRMS (EI) calcd for $C_{19}H_{32}NO_5$ (M+1) 354.22802, found 354.23059. HRMS (EI) calcd for C₁₉H₃₂NO₅ (M+1) 354.22802, found 354.23059.

Acknowledgements

Financial support by the Deutsche Forschungsgemeinschaft

and the Fonds der Chemischen Industrie is gratefully acknowledged.

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